

Phenyl Substituted Porphyrins. 6. Acylation of Alcohols with Carboxyphenylporphyrins

Mariya A. Salnikova,^a Tatiana V. Lubimova,^b Sergei A. Syrbu,^a and Alexander S. Semeikin^{a@}

^aResearch Institute of Macroheterocycles, Ivanovo State University of Chemistry and Technology, 153000 Ivanovo, Russia

^bInstitute of Solution Chemistry, Russian Academy of Sciences, 153045 Ivanovo, Russia

@Corresponding author E-mail: semeikin@isuct.ru

Pyrrolic precursors and benzaldehydes were used to afford meso-carboxyphenylporphyrins bearing one or four carboxyl groups. The reaction conditions for the acylation of alcohols with the obtained carboxyphenylporphyrins in the presence of carbodiimides were studied. Optimal reaction conditions were found and a series of phenylporphyrin esters of various structure was synthesized.

Keywords: Porphyrins, acylation, perfluorinated alcohols, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDAC), 4-dimethylaminopyridine (DMAP).

Фенилзамещенные порфирины. 6. Ацилирование спиртов карбоксифенилпорфиринами

М. А. Сальникова,^a Т. В. Любимова,^b С. А. Сырбу,^a А. С. Семейкин^{a@}

^aНИИ Макрогетероциклических соединений, Ивановский государственный химико-технологический университет, 153000 Иваново, Россия

^bИнститут химии растворов РАН, 153045 Иваново, Россия

@E-mail: semeikin@isuct.ru

Из пиррольных предшественников и бензальдегидов синтезированы мезо-карбоксифенилпорфирины, содержащие одну или четыре карбоксильные группы. Изучены условия ацилирования спиртов полученными карбоксифенилпорфиринами в присутствии карбодиимидов. Найдены оптимальные условия проведения реакции, и синтезирован ряд сложных эфиров фенилпорфиринов различной структуры.

Ключевые слова: Порфирины, ацилирование, перфторированные спирты, 1-(3-диметиламинопропил)-3-этилкарбодиимид гидрохлорид (EDAC), 4-диметиламинопиридин (DMAP).

Introduction

meso-Substituted porphyrins bearing active groups in phenyl substituents present a considerable interest in the aspect of using them as basis for more complex conjugates in order to study the influence of substituents on the properties of the porphyrin core.

Previously we have described the procedures for acylation of hydroxy^[1] and aminophenyl^[2] porphyrins with carbonic acids and the synthesis of tetraphenylporphyrin

fluoroalkyl derivatives.^[3] Here we present the study of the specifics of alcohols acylation with carboxyphenyl porphyrins resulting in the formation of esters.

Various methods of acylation of amino^[4-11] and hydroxyl compounds^[12-15] with carboxyphenylporphyrins have been published. The acylation agents can be either the carboxyphenylporphyrins themselves in the presence of activators,^[8-11,13-15] or the corresponding chloroanhydrides.^[4-7,12] Acylation is conducted in methylene chloride,^[6,9,10,13,14] chloroform,^[7,8] THF,^[15] DMF^[11] and other solvents with

pyridine added.^[4,5,12] DCC^[13,15] or EDAC^[9,10,14,15] in the presence of DMAP are used as the condensing agents.

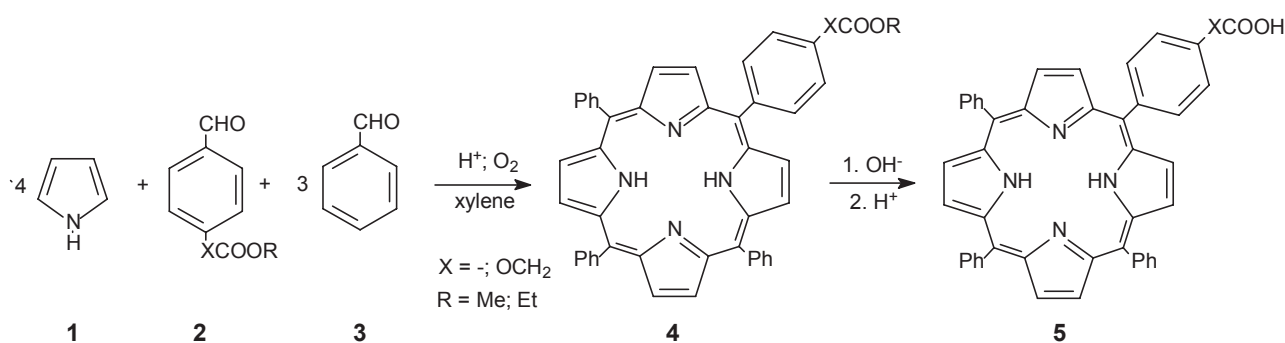
Results and Discussion

Firstly, we obtained a series of carboxyphenyl porphyrins of various nature, such as tetraphenylporphyrin derivatives bearing one (**5**, **9**) or four carboxyl groups (**11**), as well as several β -octaalkyl substituted porphyrins (**14**). Carboxyphenylporphyrins [**5** (X=–), **5** (X=OCH₂), **9**] were obtained in the course of a “mixed aldehyde condensation”. The presence of the carboxyl group does not allow the separation by chromatography, so the synthesis was conducted using the corresponding esters of carboxybenzaldehydes (**2**, **6**) with the subsequent hydrolysis of the resulting esters [**4**

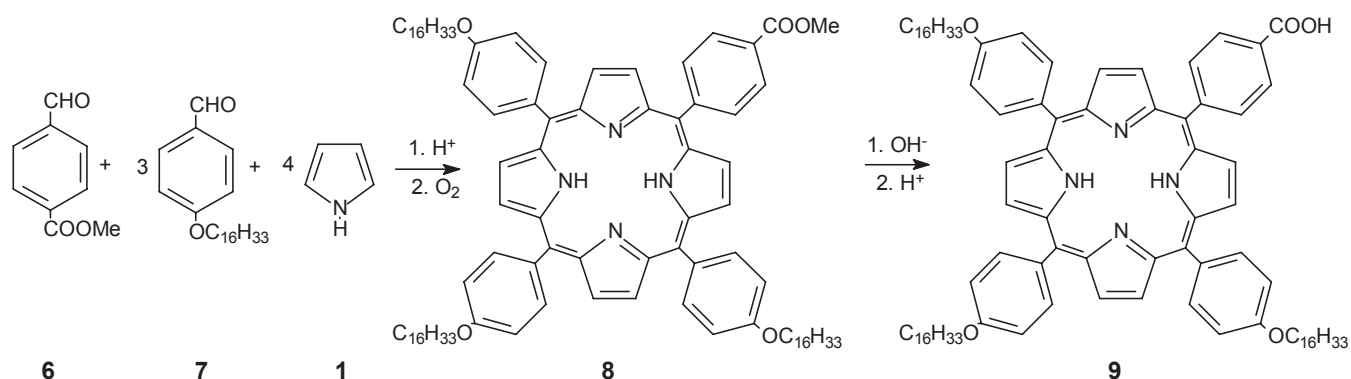
(X=–; R=Me), 8.1 % yield; **4** (X=OCH₂; R=Et), 10 % yield and **8**, 9.6% yield] (Schemes 1, 2).

Via the condensation of 3- and 4-carboxybenzaldehydes **10** with pyrrole **1** in a mixture of propionic acid with nitrobenzene, symmetrically substituted 5,10,15,20-tetrakis(3- and 4-carboxyphenyl)porphyrins (**11**) were obtained with the yields of 55.6 and 43.0 %, correspondingly (Scheme 3).

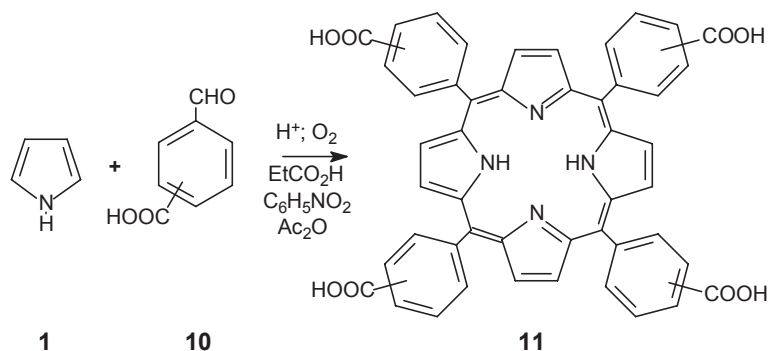
A two-step one-pot procedure analogous to^[16–18] was employed to afford carboxyporphyrins **14** by the condensation of biladiene **12** with benzaldehydes **2** (X=–; R=H and X=OCH₂; R=Et) in butanol and further hydrolysis of the obtained esters **13** (X=–; R=Me and X=OCH₂; R=Bu). The yields were 20.8 % and 45.6 %, correspondingly (Scheme 4). It should be noted that in the course of the synthesis of the ester of carboxymethyleneoxyphenylporphyrin, interesterefication occurs and butyl ester **13** (X=OCH₂; R=Bu) is obtained



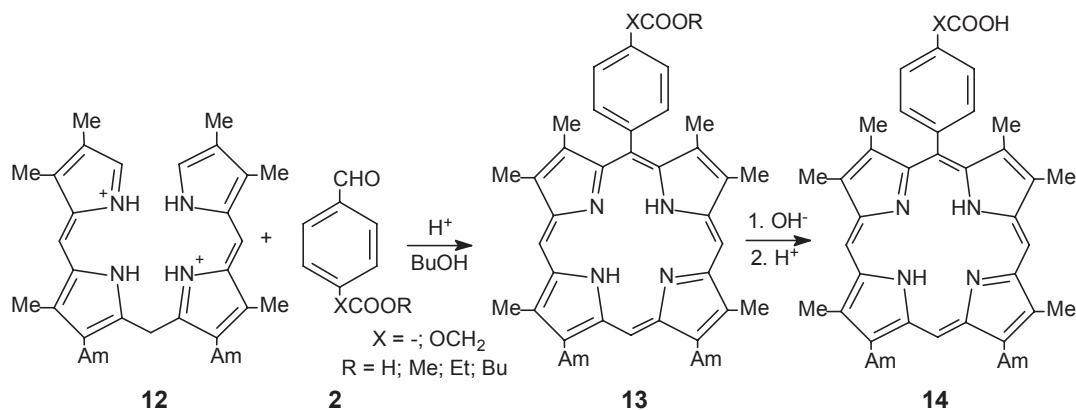
Scheme 1.



Scheme 2.



Scheme 3.



Scheme 4.

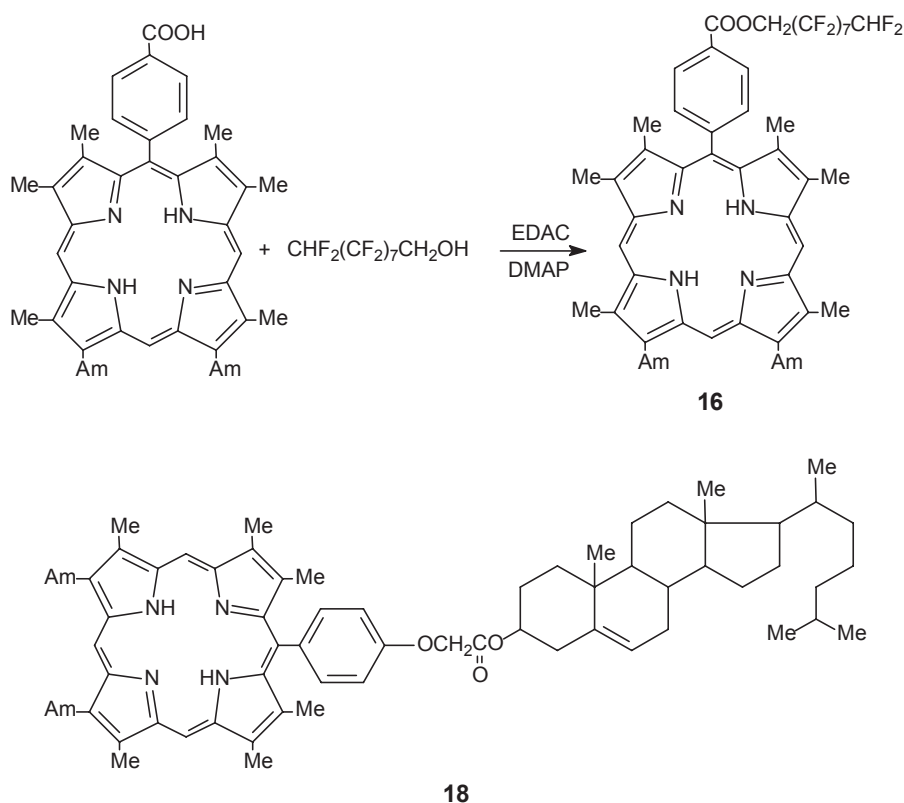
instead of the methyl one. When benzaldehyde **2** is used (X = -; R = H), esterification is insignificant and porphyrin **13** (X = -; R = Bu) is formed with 3.5 % yield. However, for chromatographic purification the porphyrin **13** (X = -; R = H) was converted into its methyl ester **13** (X = -; R = Me). The obtained esters were then subjected to hydrolysis with almost quantitative yields of the target carboxyporphyrins **14**.

Acylation of alcohols with carboxyphenylporphyrins was carried out by a previously developed procedure,^[1,2] with porphyrins being the acylating agents.

The specifics of acylation of *n*-perfluoro-1,1,9-*H*-nonyl alcohol with 5-(4-carboxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amylporphyrin **14** (X = -), resulting in the formation of porphyrin **16** (Scheme 5) (Table 1) was studied. The reaction conditions appeared to be the same for

Table 1. Dependency of the yield of porphyrin **16** on the amount of *n*-perfluoro-1,1,9-*H*-nonyl alcohol, EDAC and DMAP.

Alcohol (eq)	EDAC (eq)	DMAP (eq)	Yield, %
1.1	2	0.6	50.8
1.1	1.5	0.6	52.5
1.5	1.5	0.6	54.6
1.5	2	0.6	66.1
1.5	1.1	0.6	33.5
2	2	0.6	64.4
1.5	1.5	1	47.2
1.5	2	1	57.5
1.5	2.5	0.6	65.7



Scheme 5.

the acylation of aminophenylporphyrins with carbonic acids.^[2] Acylation of cholesterol with porphyrin **14** ($X=\text{OCH}_2$) afforded a conjugate **18** (12 % yield).

This method was also used to convert **5** ($X=-$) and **9** into porphyrins **17**, bearing polyfluorinated substituents (Scheme 6). The yields were 63 and 45 %, correspondingly.

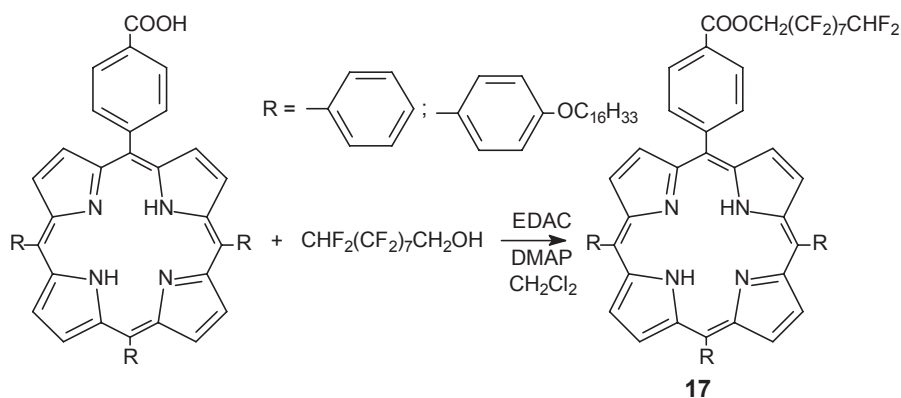
Porphyrins that may exhibit the properties of liquid crystals were synthesized. In three steps we obtained a porphyrin-cholesterol conjugate **19**, which has a discotic and cholesteric residues linked by a long decamethylene spacer (Scheme 7). Alkylation of oxyphenylporphyrin **20** with methyl ester of 11-bromoundecanoic acid gives the ester **21** (84 % yield), which after alkaline hydrolysis affords carboxyporphyrin **22** (67 % yield) and the reaction

with cholesterol results in the target conjugate **19**. The yield was 32 %, while using the corresponding chloroanhydride afforded only 13 % yield.

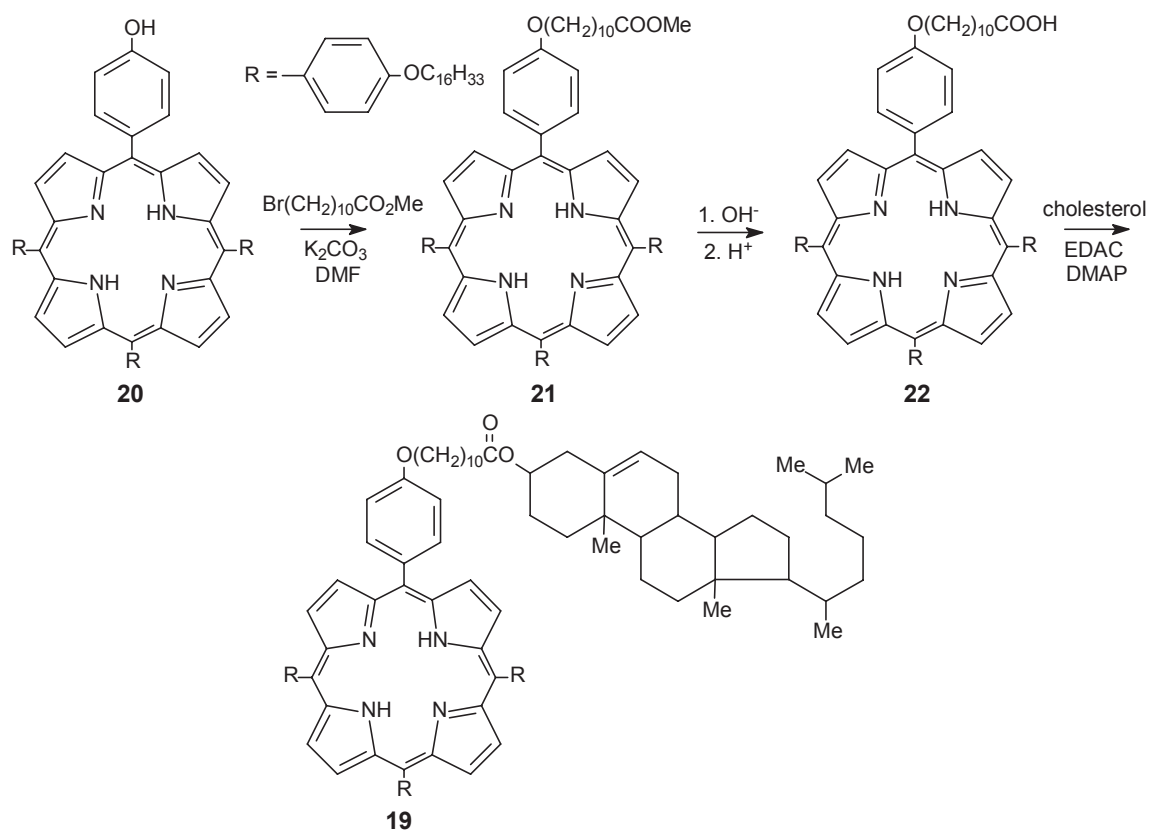
In a similar way, the reaction of **22** with aminophenylporphyrin **23** was used to obtain a dimeric porphyrin **24** with 35 % yield (Scheme 8).

While the symmetric compound 5,10,15,20-tetrakis-(4-hexadecyloxyphenyl)porphine exhibits a monotropic liquid crystal phase that vitrifies preserving the *meso*-phase texture, polarization microscopy studies of the mesomorphic properties of unsymmetrical analogs showed that compound **20** is not mesomorphic.^[3]

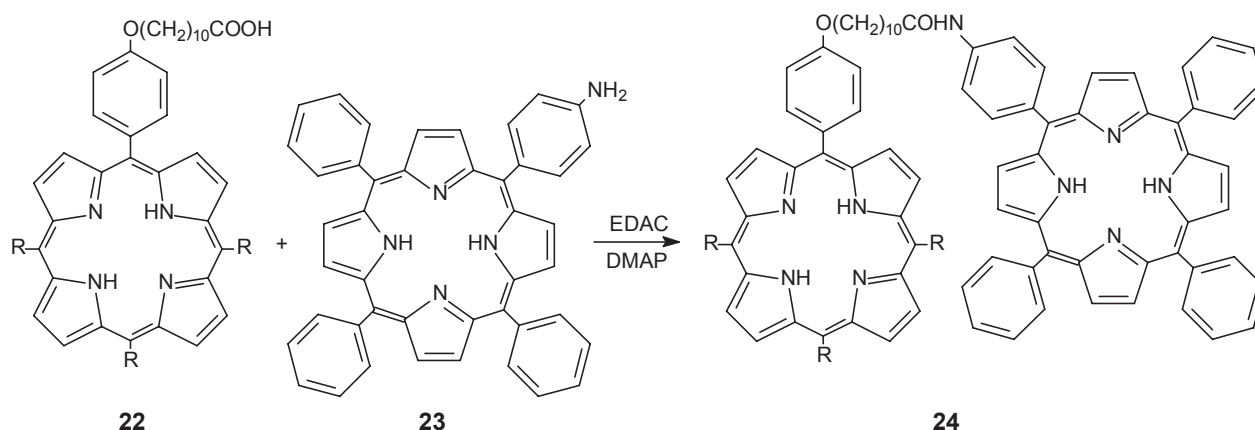
In order to study the discotic liquid crystal properties of porphyrins we synthesized symmetric porphyrins



Scheme 6.



Scheme 7.



Scheme 8.

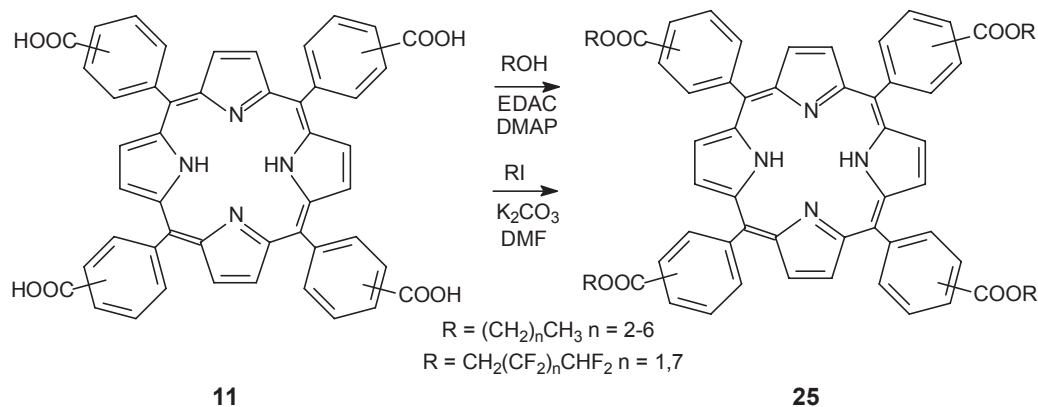
bearing ester groups with long-chain substituents of different nature (aliphatic and perfluorinated residues) **25**, by the acylation of the corresponding alcohols with *meso*-tetracarboxyphenylporphyrins **11** (Scheme 9) (Table 2).

The low yields of porphyrins **25** are caused, to our suggestion, by the complete insolubility of the starting carboxyphenylporphyrins in methylene chloride that hinders the formation of the intermediate *O*-acylureas. This assumption is confirmed by increased (by ~6 %) yields when preliminary ultrasonic treatment of the starting compounds is performed. Moreover, the presence of four carboxyl groups

leads to the formation of porphyrinic mixtures in which only the tetrasubstituted porphyrins are chromatographically mobile.^[3]

Furthermore we conducted the synthesis of tetracarboxyporphyrin esters **25** by nucleophilic substitution of halogen in alkyl halides in DMF in the presence of potash, however the yields in this case are also low and the reaction proceeds slowly (Scheme 9).

All the obtained porphyrins were analyzed using a complex of spectral studies (UV-vis, IR and ¹H NMR spectroscopy) and their purity and identity were confirmed by TLC.



Scheme 9.

Table 2. Yields of the symmetrically substituted esters **25**.

R = -(CH ₂) _n CH ₃			R = -CH ₂ (CF ₂) _n CHF ₂		
Position of the substituent	n	Yield, %	Position of the substituent	n	Yield, %
<i>para</i> -	2	26.2	<i>para</i> -	1	64.8
<i>para</i> -	3	26.2			27.2
<i>para</i> -	5	23.2 (23.1)*	<i>para</i> -	7	33.0 (US)
<i>para</i> -	6	22.4			
<i>para</i> -	15	11.4 (21.3)*	<i>meta</i> -	7	25.2 (US)

* – synthesis via alkyl halides; US – ultrasonic treatment.

Experimental

The electronic absorption spectra (EAS) were recorded on a SPEC SSP-715 scanning spectrometer in chloroform, the IR spectra were taken on an Avatar 360 FT-IR in KBr pellets, the ^1H NMR spectra were recorded on a Bruker 500 MHz instrument in deuterated chloroform (TMS as the internal standard) in the joint research center "The Upper-Volga Regional Center of Physico-Chemical Studies" of the G.A. Krestov Institute of Solution Chemistry of the Russian Academy of Sciences. The TLC was performed on Silufol plates. The solvents used were dried and purified by standard procedures.^[19]

4-Ethoxycarbonylmethyleneoxybenzaldehyde (2, $X=\text{OCH}_2$, $R=\text{Et}$). 5.0 g (40.9 mmol) of 4-oxybenzaldehyde, 9.0 mL (85.1 mmol) of ethyl ester of chloroacetic acid, 3.0 g (21.7 mmol) of potassium carbonate in DMF (75 mL) were refluxed with stirring for 1 h. The mixture was then cooled, poured into 200 mL of water and extracted with ether. The ethereal solution was washed with water and dried with sodium sulfate. Ether was evaporated off and the residue was distilled under the vacuum of a water jet pump, the fraction boiling at $\sim 185^\circ\text{C}$ was collected. Yield: 7.3 g (86 %). ^1H NMR (CCl_4) δ ppm: 9.57 s (1H, CHO); 7.57 d (2H, $J=8.0$ Hz, 2,6-H); 6.72 d (2H, $J=8.0$ Hz, 3,5-H); 4.45 s (2H, OCH_2CO); 4.00 q (2H, $^1J=7.1$ Hz, $\text{OCH}_2\text{-Et}$); 1.15 t (3H, $^1J=7.1$ Hz, $\text{CH}_3\text{-OEt}$).

4-Methoxycarbonylbenzaldehyde (2, $X=-$, $R=\text{Me}$). To a stirred suspension of 4-carboxybenzaldehyde (12.0 g, 0.08 mol) and potash (35.0 g, 0.25 mol) in 250 mL of acetone methyl iodide (5.5 mL, 0.088 mol) was added. The mixture was refluxed with stirring for 4 h, the residue was filtered off, washed with acetone and the acetone was then evaporated. The residue was recrystallized from methanol. Yield: 11.6 g (88 %), m.p. $60-62^\circ\text{C}$. ^1H NMR (CCl_4) δ ppm: 9.53 s (1H, CHO); 7.72 d (2H, $J=8.1$ Hz, 3,5-H); 7.57 d (2H, $J=8.1$ Hz, 2,6-H); 4.30 s (3H, OCH_3).

4-Hexadecyloxybenzaldehyde (7). A mixture of 20.0 g (0.164 mol) of 4-oxybenzaldehyde, 50 mL (0.164 mol) of cetyl bromide, 25.0 g (0.181 mol) of potassium carbonate and 200 mL of *n*-butanol was refluxed with stirring for 3 h. The mixture was then cooled, the precipitate was filtered, washed with water and air-dried at ambient temperature. Yield: 43.9 g (77 %), m.p. $41-42^\circ\text{C}$. ^1H NMR (CCl_4) δ ppm: 9.89 s (1H, CHO); 7.83 d (2H, $J=8.5$ Hz, 2,6-H); 7.01 d (2H, $J=8.5$ Hz, 3,5-H); 4.05 t (2H, $^1J=7.0$ Hz, OCH_2); 1.83 qv (2H, $^1J=7.0$ Hz, CH_2); 1.44-1.52 m (2H, CH_2); 1.23-1.41 m (24H, CH_2); 0.90 t (3H, $^1J=7.0$ Hz, CH_3).

5-(4-Methoxycarbonylphenyl)-10,15,20-triphenylporphyrin (4, $X=-$, $R=\text{Me}$). To a solution of 410 mg (2.5 mmol) of 4-methoxycarbonylbenzaldehyde, 0.7 mL (10 mmol) of pyrrole and 0.9 mL (8.9 mmol) of benzaldehyde in 1000 mL of methylene chloride 1.0 g of finely powdered ammonium chloride was added, then in an argon atmosphere a solution of 0.3 mL (2.4 mmol) of boron trifluoride etherate in 10 mL of methylene chloride was dropwise added. The mixture was stirred for additional 1 h and then treated with 1.85 g (7.5 mmol) of *p*-chloroanil and stirred overnight. The solvent was removed, the residue washed with sodium hydroxide and water to neutral pH and then air-dried at 70°C . The mixture of porphyrins was then dissolved in methylene chloride and subjected to chromatography on aluminum oxide (Brockmann II grade), porphyrin was eluted as the second zone, after tetraphenylporphyrin. The eluate was evaporated to dryness, the residue was dissolved in benzene and subjected to chromatography on silica gel (L 100/250), using benzene as an eluent, the solvent was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and air-dried. Yield 136 mg (8.1 %). $R_f=0.63$ (benzene-methanol, 10:1). EAS λ_{max} (lge): 645 (3.72); 589 (3.88); 549 (3.97); 514 (4.35); 417 (5.59). ^1H NMR (CDCl_3) δ ppm: 8.86-8.91 m (6H, β -H); 8.82 d (2H, $J=4.5$ Hz, 3,7-H); 8.46 d (2H, $^1J=8.2$ Hz, 2,6-H-Ar); 8.33 d (2H, $^1J=8.2$ Hz, 3,5-H-Ar); 8.24 d (6H, $^2J=6.9$ Hz, 2,6-H-Ph); 7.76-7.84 m (9H, 3,4,5-H-Ph); 4.14 s (3H, OCH_3); -2.77 bs (2H, NH).

5-(4-Ethoxycarbonylmethyleneoxyphenyl)-10,15,20-triphenylporphyrin (4, $X=\text{OCH}_2$, $R=\text{Et}$). To a refluxing solution of 4 mL (53.8 mmol, $\sim 2\%$) of trifluoroacetic acid in 300 mL of *p*-xylene under argon was gradually added a solution of 3.8 g (18.05 mmol) 4-ethoxycarbonylbenzaldehyde, 5.5 mL (54.18 mmol) benzaldehyde and 5.0 mL (72.2 mmol) pyrrole in 50 mL *p*-xylene. The mixture was refluxed for 0.5 h under argon and then for 1 h under a stream of air. The solvent was removed by distillation with water vapor, the residue filtered, washed with water and dried. The mixture was dissolved in chloroform and subjected to chromatography on a column with Al_2O_3 (II Brockmann grade), eluting the first fraction of tetraphenylporphyrin with chloroform. The eluate was evaporated and the porphyrin precipitated with methanol. Yield: 1.1 g (9.9 %). The second fraction was eluted with a mixture of chloroform-methanol, evaporated to dryness and subjected to chromatography on silica gel (L 100/250) eluting with methylene chloride and collecting the second porphyrinic fraction. The eluate was evaporated and the porphyrin was precipitated with methanol. Yield: 1.3 g (10.0 %). EAS λ_{max} (lge): 650 (3.83); 594 (4.03); 554 (4.03); 518 (4.23); 422 (5.66). ^1H NMR δ ppm: 8.92-8.99 m (8H, β -H); 8.25 d (6H, $J=6.6$ Hz, 2,6-H-Ph); 8.00 d (2H, $^1J=7.6$ Hz, 2,6-H-Ar); 7.74-7.83 m (9H, 3,4,5-H-Ph); 7.19 d (2H, $^1J=7.6$ Hz, 3,5-H-Ar); 4.84 s (2H, OCH_2CO); 4.34 q (2H, $^2J=7.2$ Hz, $\text{OCH}_2\text{-Et}$); 1.34 t (3H, $^2J=7.2$ Hz, $\text{CH}_3\text{-Et}$); -2.74 bs (2H, NH).

5-(4-Methoxycarbonylphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amyloporphyrin (13, $X=-$, $R=\text{Me}$). To a solution of 1.38 g (4.39 mmol) of 4,4'-dimethyl-3,3'-diamyldipyrrromethane^[18] and 1.1 g (8.93 mmol) of 2-formyl-3,4-dimethylpyrrole^[18] in 100 mL of butanol at ambient temperature and stirring, conc. hydrobromic acid (2.0 mL, ~ 16.6 mmol) was added (biladiene was precipitated), in 1 h 2.7 g (18.0 mmol, 4-fold excess) of 4-carboxybenzaldehyde was added, the mixture was refluxed for 4 h, then 1.1 g (4.39 mmol) of iodine was added, the mixture was stirred for 15 min and chilled. Butanol was removed by steam distillation, the residue was filtered, washed with water and air-dried at 70°C . The mixture was dissolved in methylene chloride and subjected to chromatography on silica gel eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. Yield: 100 mg (3.5 %), **13** ($X=-$; $R=\text{Bu}$). The insoluble precipitate was stirred for several days at ambient temperature in a mixture of sulfuric acid (4 mL) and methanol (150 mL). Half of the methanol was removed, the rest was diluted with water and neutralized with aqueous ammonia, the precipitate was filtered, washed with water and air-dried at 70°C . It was then dissolved in methylene chloride and chromatographed on silica gel eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. Yield: 0.6 g (20.8 %), **13** ($X=-$; $R=\text{Bu}$): $R_f=0.51$ (benzene). EAS λ_{max} (lge): 623 (3.63); 571 (3.90); 537 (3.94); 504 (4.21); 404 (5.25). ^1H NMR δ ppm: 10.18 s (2H, 10,20-H); 9.97 s (1H, 15-H); 8.46 d (2H, $J=7.8$ Hz, 2,6-H-Ar); 8.21 d (2H, $J=7.8$ Hz, 3,5-H-Ar); 4.53 t (8H, $^1J=6.9$ Hz, $\text{OCH}_2\text{-Bu}$); 4.04 t (4H, $^2J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 3.62 s (6H, 12,18- CH_3); 3.51 s (6H, 2,8- CH_3); 2.48 s (6H, 3,7- CH_3); 2.32 qv (4H, $^2J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 1.94 qv (2H, $^1J=6.9$ Hz, $\text{CH}_2\text{-Bu}$); 1.75 qv (4H, $^2J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 1.52-1.46 m (6H, $\text{CH}_2\text{-Bu}+\text{CH}_2\text{-Am}$); 1.17 t (3H, $^1J=6.9$ Hz, $\text{CH}_3\text{-Bu}$); 0.99 t (6H, $^2J=7.5$ Hz, $\text{CH}_3\text{-Am}$); -3.36 bs (2H, NH). **13** ($X=-$; $R=\text{Me}$): $R_f=0.42$ (benzene). EAS λ_{max} (lge): 624 (3.83); 571 (3.88); 538 (3.91); 504 (4.18); 405 (5.23). ^1H NMR δ ppm: 10.19 s (2H, 10,20-H); 9.95 s (1H, 15-H); 8.43 d (2H, $J=7.8$ Hz, 2,6-H-Ar); 8.21 d (2H, $J=7.8$ Hz, 3,5-H-Ar); 4.35 s (3H, OCH_3); 4.05 t (4H, $^1J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 3.62 s (6H, 12,18- CH_3); 3.54 s (6H, 2,8- CH_3); 2.50 s (6H, 3,7- CH_3); 2.33 qv (4H, $^1J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 1.75 qv (4H, $^1J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 1.55 s (4H, $^1J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 0.99 t (6H, $^1J=7.5$ Hz, $\text{CH}_3\text{-Am}$); -3.34 bs (2H, NH).

5-(4-Butoxycarbonylmethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amyloporphyrin (13, $X=\text{OCH}_2$, $R=\text{Bu}$). To a solution of 0.67 g (2.13 mmol) 4,4'-dimethyl-3,3'-

diamyldipyrromethane and 0.52 g (4.26 mmol) of 2-formyl-3,4-dimethylpyrrole in 50 mL of *n*-butanol at ambient temperature, concentrated hydrobromic acid (1.0 mL, 8.3 mmol) was added with stirring. The mixture was stirred for 1 h, then 3.0 mL (~6-fold excess) of 4-ethoxycarbonylmethyleneoxybenzaldehyde was added refluxed for 4 h. After cooling, 2 mL of conc. aqueous ammonia was introduced. The precipitate was filtered off, washed with methanol and air-dried. Then it was dissolved in chloroform and purified by column chromatography on silica gel (L 100/250), using chloroform-methanol mixture (100:1) as the eluent. Yield: 0.72 g (45.6 %). $R_f=0.86$ (benzene:methanol=20:1). EAS λ_{\max} (lge): 623.5 (3.62); 571 (3.95); 537 (3.97); 503 (4.27); 404.2 (5.36). IR (KBr) cm^{-1} : $\nu_{\text{CO}} - 1633$. $^1\text{H NMR}$ δ ppm: 10.16 s (2H, 10.20-H); 9.96 s (1H, 15-H); 7.93 d (2H, $J=8.0$ Hz, 2.6-H-Ar); 7.27 d (2H, $J=8.0$ Hz, 3.5-H-Ar); 4.95 s (2H, OCH_2CO); 4.39 t (2H, $^1J=7.7$ Hz, $\text{OCH}_2\text{-Bu}$); 4.05 t (4H, $^2J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 3.65 s (6H, 12.18- CH_3); 3.55 s (6H, 2.8- CH_3); 2.50 s (6H, 3.7- CH_3); 2.33 qv (4H, $^2J=7.5$ Hz, $\text{CH}_2\text{-Am}$); 1.71-1.78 m (6H, $\text{CH}_2\text{-Bu}$, Am); 1.52-1.60 m (6H, $\text{CH}_2\text{-Bu}$, Am); 1.06 t (3H, $^1J=7.7$ Hz, $\text{CH}_3\text{-Bu}$); 0.99 t (6H, $^2J=7.5$ Hz, $\text{CH}_3\text{-Am}$); -3.16 bs (1H, NH); -3.33 bs (1H, NH).

5-(4-Carboxyphenyl)-10,15,20-triphenylporphyrine (5, $X=-$). 100 mg (0.15 mmol) of 5-(4-methoxycarbonyl)phenyl-10,15,20-triphenylporphyrine, and 100 mg (1.8 mmol) of potassium hydroxide was dissolved in a mixture of propyl alcohol (20 mL), THF (20 mL) and water (2.5 mL) and refluxed for 3 h. The solution was then cooled and treated with dilute hydrochloric acid until the color changed from crimson to green. The formed precipitate was filtered, washed with water, dissolved in a mixture of chloroform-methanol and washed with water in a separatory funnel until the color changed from green to red. The solution was dried with sodium sulfate, chloroform was evaporated and the residue was air-dried. Yield: 80 mg (81.8 %). EAS λ_{\max} (lge): 646 (3.67); 590 (3.81); 551 (3.94); 516 (4.25); 419 (5.63).

Similarly obtained were:

5-(4-Carboxymethyleneoxyphenyl)-10,15,20-triphenylporphyrine (5, $X=\text{OCH}_2$). Yield: 97.5 %. EAS λ_{\max} (lge): 649 (3.81); 594 (4.00); 553 (4.05); 518 (4.19); 420 (5.63).

5-(4-Carboxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amylporphyrine (14, $X=-$). Yield: 89.5 %. EAS λ_{\max} nm (lge): 646 (3.65); 590 (3.78); 550 (3.91); 515 (4.25); 419 (5.62).

5-(4-Carboxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphyrine (9). To a refluxing solution of 2 mL (26.9 mmol) of trifluoroacetic acid in 300 mL of *p*-xylene in an argon atmosphere it was gradually added a solution of 3.0 g (18.27 mmol) of 4-methoxycarbonylbenzaldehyde **6**, 18.9 g (54.5 mmol) of 4-hexadecyloxybenzaldehyde **7** and 5.0 mL (72.2 mmol) of pyrrole in *p*-xylene (50 mL). The mixture was refluxed for 0.5 h in the inert atmosphere and then for 1 h under a stream of air. The solvent was then steam distilled, the residue filtered and dried at 70 °C. The porphyrin mixture was dissolved in methylene chloride and chromatographed on silica gel (L 100/250), using methylene chloride as the eluent. A secondary purification run was performed on Al_2O_3 (II Brockmann grade), using benzene as the eluent. The second porphyrin band was collected, the eluent evaporated and the porphyrin was precipitated with methanol, filtered and air-dried. The obtained 5-(4-methoxycarbonyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphyrine **8** was refluxed for 4 h in a solution of KOH (5.0 g, 89.1 mmol) in 10 mL of water and 150 mL of propyl alcohol. The reaction mixture was then treated with water, the formed precipitate was filtered, washed with water and dried. Yield of the potassium salt: 3.2 g. The salt was dissolved in methylene chloride and shaken with 5 % hydrochloric acid. The organic layer was then washed with water until the color changed from green to red. The solution was evaporated and the porphyrin was precipitated with methanol, filtered and air-dried at ambient temperature. Yield: 2.4 g (9.6 %). EAS λ_{\max} (lge): 649 (3.83); 593 (3.82); 555 (4.00); 518 (4.16); 421 (5.43). $^1\text{H NMR}$ δ ppm: 8.88-8.97 m (6H, β -H); 8.78-8.84 m (2H, β -H); 8.46-8.54 m (2H, 2.6-H-Ar); 8.31-8.39 m (2H, 3.5-H-Ar); 8.09-8.17 m

(6H, 2.6-H-Ar'); 7.26-7.34 m (6H, 3.5-H-Ar'); 4.22-4.32 m (6H, OCH_2); 1.96-2.07 m (6H, CH_2); 1.61-1.70 m (6H, CH_2); 1.56-1.12 m (72H, CH_2); 0.85-0.96 m (9H, CH_3); -2.72 bs (2H, NH).

5,10,15,20-Tetrakis(4-carboxyphenyl)porphyrine (11, *para*). The solution of 0.7 mL (10.1 mmol) of pyrrole **1**, 1.5 g (10 mmol) of 4-carboxybenzaldehyde **10** (*para*) and 1.7 mL (18.0 mmol) of acetic anhydride in a mixture of nitrobenzene (15 mL) and propionic acid (35 mL) was heated and refluxed for 2 h. The reaction mixture was then maintained at room temperature overnight. The porphyrin precipitate was filtered, washed with methanol and air-dried at 70 °C. Yield: 1.1 g (55.6 %). EAS (2 % KOH in water) λ_{\max} (lge): 651 (3.88); 600 flat (3.92); 565 (4.04); 524 (4.10); 413 (5.15).

Similarly obtained was:

5,10,15,20-Tetrakis(3-carboxyphenyl)porphyrine (11, *meta*). Yield: 43.0 %. EAS (2% KOH in water) λ_{\max} (lge): 639 (3.66); 581 (3.78); 554 (3.85); 518 (4.09); 414 (5.49).

5-(4-Carboxymethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-diamylporphyrine (14, $X=\text{OCH}_2$). A solution of 0.44 g (0.59 mmol) of 5-(4-butoxycarbonylmethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amylporphyrine **13** ($X=\text{OCH}_2$; $R=\text{Bu}$) in 50 mL of methanol in the presence of KOH (1.0 g) in 10 mL of water was refluxed for 24 h. The formed precipitate was filtered, washed with methanol and dried. 290 mg (0.37 mmol) of the obtained potassium salt of 5-(4-carboxymethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amylporphyrine was treated with a solution of 1 mL of 35 % hydrochloric acid in 5 mL of methanol, then water (20 mL) was introduced. The formed precipitate of the carbonic acid was filtered, washed with water and air-dried at 70 °C. Yield: 250 mg (~100 %). EAS λ_{\max} (lge): 621 (3.52); 569.5 (3.87); 537 (3.88); 503 (4.18); 402.6 (5.29).

5-[4-(1,1,9-*H*-Perfluorononyloxycarbonyl)phenyl]-2,3,7,8,12,18-hexamethyl-13,17-di-*n*-amylporphyrine (16). A mixture of 50 mg (0.073 mmol) of 5-(4-carboxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-diamylporphyrine **14** ($X=-$), 50 mg (0.116 mmol) of *n*-perfluoro-1,1,9-*H*-nonyl alcohol, 6 mg (0.049 mmol) of DMAP and 30 mg (0.156 mmol) of EDAC was dissolved in 15 mL methylene chloride and stirred at 0 °C for 1.5 h, then at ambient temperature until the reaction was complete according to TLC (5 days). The reaction mixture was purified by chromatography on silica gel (L 100/250), using methylene chloride as the eluent. The eluate was evaporated and the porphyrin was precipitated with methanol. The residue was filtered, washed with methanol and dried. Yield: 53.8 mg (66.1 %). $R_f=0.67$ (benzene). EAS λ_{\max} nm (lge): 624 (3.71); 581 (3.86); 538 (3.99); 504 (4.25); 405 (5.32). IR cm^{-1} : $\nu_{\text{OCO}} - 1737$; $\nu_{\text{CF}} - 1213$. $^1\text{H NMR}$ δ ppm: 10.18 s (2H, 10.20-H); 9.97 s (1H, 15-H); 8.46 d (2H, $J=7.8$ Hz, 2.6-H-Ar); 8.21 d (2H, $J=7.8$ Hz, 3.5-H-Ar); 6.10 tt (1H, $^1J=5.1$ Hz, $^2J=51.9$ Hz, CHF_2); 5.07 t (2H, $^3J=13.4$ Hz, OCH_2); 4.04 t (4H, $^4J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 3.64 s (6H, 12.18- CH_3); 3.54 s (6H, 2.8- CH_3); 2.43 s (6H, 3.7- CH_3); 2.33 qv (4H, $^4J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 1.76 qv (4H, $^4J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 1.57 sc (4H, $^4J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 1.00 t (6H, $^4J=7.6$ Hz, $\text{CH}_3\text{-Am}$); -2.96 bs (2H, NH).

5-[4-(1,1,9-*H*-Perfluorononyloxycarbonyl)phenyl]-10,15,20-triphenylporphyrine (17, $R=\text{Ph}$). A mixture of 50 mg (0.073 mmol) of 5-(4-carboxyphenyl)-10,15,20-triphenylporphyrine **5** ($X=-$), 49 mg (0.113 mmol) of *n*-perfluoro-1,1,9-*H*-nonyl alcohol, 6 mg (0.049 mmol) of DMAP and 30 mg (0.156 mmol) of EDAC was dissolved in 15 mL of methylene chloride and stirred at 0 °C for 1.5 h, then at ambient temperature until the reaction was complete according to TLC (1 day). The reaction mixture was purified by chromatography on silica gel (L 100/250), using methylene chloride as the eluent. The eluate was evaporated and the porphyrin was precipitated with methanol. The residue was filtered, washed with methanol and dried. Yield: 51.1 mg (62.9 %). $R_f=0.75$ (benzene). EAS λ_{\max} nm (lge): 646 (3.76); 590 (3.89); 553 (4.00); 516 (4.32); 419 (5.68). IR cm^{-1} : $\nu_{\text{OCO}} - 1733$; $\nu_{\text{CF}} - 1210$. $^1\text{H NMR}$ δ ppm: 8.90 d (2H, $J=4.6$ Hz, 2.8-H); 8.89 s (4H, 12.13.17.18-H); 8.82 d (2H, $J=4.6$ Hz, 3.7-H); 8.49 d (2H,

$^1J=8.0$ Hz, 2.6-H-Ar); 8.39 d (2H, $^1J=8.0$ Hz, 3.5-H-Ar); 8.24 d (6H, $^2J=7.0$ Hz, 2.6-H-Ph); 7.75-7.85 m (9H, 3.4.5-H-Ph); 6.10 tt (1H, $^3J=5.1$ Hz, $^4J=51.9$ Hz, CHF_2); 5.05 t (2H, $^5J=13.3$ Hz, OCH_2); -2.75 bs (2H, NH).

5-[4-(1,1,9-H-Perfluorononyloxy)carbonylphenyl]-10,15,20-tris(4-n-hexadecyloxy-phenyl)porphine (17), $R=4\text{-C}_6\text{H}_4\text{OC}_{16}\text{H}_{33}$. A mixture of 52 mg (0.038 mmol) of 5-(4-carboxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphine **9**, 33.5 mg (0.078 mmol) of *n*-perfluoro-1,1,9-H-nonyl alcohol, 6.5 mg (0.053 mmol) of DMAP and 15 mg (0.070 mmol) of EDAC was dissolved in 15 mL of anhydrous methylene chloride and stirred at 0 °C for 1.5 h, then at ambient temperature until the completion of the reaction (4 days, TLC). The reaction mixture was purified by chromatography on silica gel (L 100/250), using methylene chloride as the eluent. The eluate was evaporated and the porphyrin was precipitated with methanol. The residue was filtered, washed with methanol and dried. Yield: 30.5 mg (44.7 %). $R_f=0.83$ (benzene). EAS λ_{max} nm (lge): 651 (3.53); 593 (3.85); 556 (3.92); 519 (4.16); 423 (5.58). IR cm^{-1} : $\nu_{\text{OCO}} - 1738$; $\nu_{\text{CF}} - 1211$. $^1\text{H NMR}$ δ ppm: 8.94 d (2H, $J=4.2$ Hz, 2.8-H); 8.92 s (4H, 12.13.17.18-H); 8.80 d (2H, $J=4.2$ Hz, 3.7-H); 8.50 d (2H, $^1J=8.1$ Hz, 2.6-H-Ar); 8.40 d (2H, $^1J=8.1$ Hz, 3.5-H-Ar); 8.15 d (6H, $^2J=8.5$ Hz, 2.6-H-Ar); 7.32 d (6H, $^2J=8.5$ Hz, 3.5-H-Ar); 6.12 tt (1H, $^3J=51.9$ Hz, $^4J=5.0$ Hz, CHF_2); 5.06 t (2H, $^5J=13.3$ Hz, OCH_2); 4.28 t (6H, $^6J=7.0$ Hz, OCH_2); 2.02 qv (6H, $^6J=7.0$ Hz, CH_2); 1.66 qv (12H, $^6J=7.0$ Hz, CH_2); 1.52 qv (6H, $^6J=7.0$ Hz, CH_2); 1.45 qv (6H, $^6J=7.0$ Hz, CH_2); 1.26-1.41 m (54H, CH_2); 0.92 t (9H, $^6J=7.0$ Hz, CH_3); -2.71 bs (2H, NH).

5-(4-O-Cholesterylcarboxymethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-di-n-amyloporphine (18). A solution of 50.0 mg of (0.073 mmol) 5-(4-carboxymethyleneoxyphenyl)-2,3,7,8,12,18-hexamethyl-13,17-diamyloporphine **14** ($X=\text{OCH}_2$), 45 mg (0.368 mmol) of DMAP and 45 mg (0.218 mmol) of DCC in 15 mL THF was stirred at 0 °C for 1 h, then 275 mg (0.711 mmol) of cholesterol was added and the stirring was maintained for another 30 minutes. The mixture was maintained at ambient temperature for a day. The porphyrin was purified on Al_2O_3 (II Brockmann grade), eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 9.1 mg (11.8 %). $R_f=0.82$ (benzene:methanol, 20:1). EAS λ_{max} (lge): 624 (3.63); 571 (3.95); 537 (3.97); 503 (4.27); 404 (5.36). IR cm^{-1} : 1660; 1710. $^1\text{H NMR}$ (CDCl_3) δ ppm: 10.17 s (2H, 10.20-H); 9.97 s (1H, 15-H); 7.94 d (2H, $J=7.8$ Hz, 2.6-H-Ar); 7.28 d (2H, $J=7.8$ Hz, 3.5-H-Ar); 5.03 s (2H, OCH_2); 4.06 t (4H, $^1J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 3.65 s (6H, 12.18- CH_3); 3.55 s (6H, 2.8- CH_3); 2.50 s (6H, 3.7- CH_3); 2.32 qv (4H, $^1J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 1.75 qv (4H, $^1J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 1.58 sc (4H, $^1J=7.6$ Hz, $\text{CH}_2\text{-Am}$); 0.98 t (6H, $^1J=7.6$ Hz, $\text{CH}_2\text{-Am}$); -3.23 bs (2H, NH); 1.20-2.15 (a number of cholesterol residue signals).

5-[4-(10-Methoxycarbonyldecamethyleneoxy)phenyl]-10,15,20-tris(4-hexadecyloxyphenyl)porphine (21). A mixture of 500 mg (0.37 mmol) 5-(4-oxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphine **20**,^[1] 0.5 g (1.79 mmol) of methyl ester of 11-bromoundecanoic acid and 2.0 g (14.46 mmol) of potassium carbonate in 50 mL of DMF was refluxed for 5 h, then cooled, poured into 200 mL of water, the precipitate was filtered, washed with water and dried at ambient temperature in air. The obtained porphyrin was dissolved in chloroform and purified on II Brockmann grade aluminum oxide, eluting with chloroform. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 0.48 g (83.7 %). $R_f=0.54$ (benzene). EAS λ_{max} (lge): 650 (3.98); 593 (3.98); 556 (4.19); 519 (4.31); 422 (5.69). IR cm^{-1} : $\nu_{\text{OCO}} - 1740$. $^1\text{H NMR}$ δ ppm: 8.90 s (8H, β -H); 8.13 d (8H, $J=8.0$ Hz, 2.6-H-Ar); 7.30 d (8H, $J=8.0$ Hz, 3.5-H-Ar); 4.27 t (8H, $^1J=6.4$ Hz, OCH_2); 3.71 s (3H, OCH_3); 2.37 t (2H, $^2J=8.0$ Hz, COCH_2); 1.97-2.07 m (8H, CH_2); 1.61-1.73 m (10H, CH_2); 1.55-

1.47 m (8H, CH_2); 1.25-1.47 m (82H, CH_2); 0.92 t (9H, $^1J=6.4$ Hz, CH_3); -2.71 bs (2H, NH).

5-[4-(10-Carboxymethyleneoxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphine (22). A mixture of 500 mg (0.323 mmol) of 5-[4-(10-methoxycarbonyldecamethyleneoxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphine **21** and a solution of 0.1 g (1.78 mmol) of KOH in 1 mL of water and 30 mL of *n*-propanol was refluxed for 1 h. The mixture was cooled, the precipitate was filtered, washed with propanol and stirred in a mixture of methylene chloride with water and hydrochloric acid until the complete dissolution. The organic layer was washed with water until the color changed from green to red, methylene chloride was evaporated to a minimal volume and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 330 mg (66.5 %). $R_f=0.91$ (benzene:methanol, 10:1), 0.7 (benzene:methanol, 20:1). EAS λ_{max} (lge): 651 (3.99); 594 (3.97); 557 (4.20); 519 (4.32); 422 (5.72). IR cm^{-1} : $\nu_{\text{OCO}} - 1705$. $^1\text{H NMR}$ δ ppm: 8.87 s (8H, β -H); 8.06-8.15 m (8H, 2.6-H-Ar); 7.21-7.28 m (8H, 3.5-H-Ar); 4.15-4.27 m (8H, OCH_2); 2.35-2.44 m (2H, COCH_2); 1.92-2.04 m (8H, CH_2); 1.67-1.74 m (2H, CH_2); 1.57-1.67 m (8H, CH_2); 1.22-1.54 m (82H, CH_2); 0.86-0.96 m (9H, CH_3); -2.72 bs (2H, NH).

5-[4-(10-O-Cholesterylcarboxymethyleneoxy)phenyl]-10,15,20-tris(4-hexadecyloxyphenyl)porphine (19). *Method A* (via chloroanhydride). To a solution of 5-[4-(10-carboxymethyleneoxyphenyl)-10,15,20-tris(4-hexadecyloxyphenyl)porphine **22** (50 mg, 0.033 mmol) in 10 mL of benzene thionyl chloride (0.3 mL, 3.7 mmol) was added and the mixture was refluxed with the protection by a calcium chloride tube for 1 h. The benzene was distilled off, another 10 mL of benzene were added and again evaporated. To the residue, cholesterol (300 mg, 0.776 mmol) was added and the mixture was dissolved in 7 mL of pyridine. The solution was stirred at room temperature for 1 day with subsequent dilution with water. The precipitate was filtered, washed with water and air-dried at 70 °C, dissolved in methylene chloride and purified on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 8.2 mg (13.4 %). *Method B* (from acid). A solution of 50 mg (0.033 mmol) of **22**, 14.5 mg (0.119 mmol) of DMAP and 25 mg (0.121 mmol) of EDAC in 25 mL of THF was stirred at 0 °C for 1 h, then 45 mg (0.166 mmol) of cholesterol was introduced and the mixture was stirred at the same temperature for 30 min. Further stirring at ambient temperature was performed for several days until the reaction was complete. The mixture was purified on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 20 mg (32.3 %). A similar synthesis in methylene chloride afforded 4.9 mg (7.8 %) of the porphyrin. $R_f=0.59$ (benzene). EAS λ_{max} (lge): 650 (3.48); 594 (3.48); 557 (3.70); 519 (3.83); 422 (5.24). IR cm^{-1} : $\nu_{\text{OCO}} - 1743$. $^1\text{H NMR}$ δ ppm: 8.89 s (8H, β -H); 8.10-8.18 m (8H, 2.6-H-Ar); 7.26-7.35 m (8H, 3.5-H-Ar); 4.33-4.44 m (8H, OCH_2); 2.43-2.54 m (2H, COCH_2); -2.74 bs (2H, NH); 0.6-2.10 (a series of resonance signals of alkyl chains and cholesterol residue).

Dimeric porphyrin (24). A solution of 50 mg (0.033 mmol) of 5-[4-(10-carboxydecamethyleneoxy)phenyl]-10,15,20-tris(4-hexadecyloxyphenyl)porphine **22**, 24 mg (0.116 mmol) of EDAC and 14 mg (0.115 mmol) of DMAP in 20 mL of methylene chloride was cooled in a water bath to 0 °C and then stirred for 1 h followed by the addition of solution of 5-(4-aminophenyl)-10,15,20-triphenylporphine **23** (24.5 mg, 0.039 mmol)^[2] in 5 mL of methylene chloride. The mixture was stirred at ambient temperature for 24 h and then refluxed while protected with a calcium chloride tube for 1 day. The dimer was purified by chromatography on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the dimeric porphyrin was precipitated with

methanol. The precipitate was filtered, washed with methanol and dried. Yield: 24.5 mg (35.1 %). $R_f=0.21$ (benzene). EAS λ_{\max} (lge): 649 (4.23); 592 (4.26); 554 (4.45); 517 (4.65); 420 (6.03). IR cm^{-1} : 1606. ^1H NMR δ ppm: 8.83–8.95 m (16H, β -H); 8.19–8.28 m (6H, 2,6-H-Ph); 8.07–8.17 m (10H, 2,6-H-Ar); 7.71–7.86 m (9H, 3,4,5-H-Ph); 7.35 bs (1H, NHCO); 7.22–7.34 m (10H, 3,5-H-Ar); 4.19–4.32 m (8H, OCH_2); 2.53 t (2H, $J=7.0$ Hz, CH_2CO); 1.88–2.08 m (10H, CH_2); 1.59–1.74 m (10H, CH_2); 1.24–1.46 m (90H, CH_2); 0.88–0.96 m (9H, CH_3); -2.70 bs, -2.74 bs (2 \times 2H, NH).

5,10,15,20-Tetrakis(4-propyloxycarbonylphenyl)porphine (25, para-, $R=-(\text{CH}_2)_3\text{CH}_3$). A mixture of 50 mg (0.063 mmol) of 5,10,15,20-tetrakis(4-carboxyphenyl)porphine **11** (*para*), 0.5 mL (6.7 mmol) of propyl alcohol, 33 mg (0.271 mmol) of DMAP and 100 mg (0.522 mmol) of EDAC was dissolved in 15 mL of methylene chloride and stirred at 0 °C in for 1.5 h, then at ambient temperature for 5 days. The porphyrin was isolated by chromatography on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 15.9 mg (26.2 %). $R_f=0.22$ (benzene). EAS λ_{\max} nm (lge): 647 (3.81); 591 (3.95); 551 (4.09); 516 (4.38); 421 (5.73). IR cm^{-1} : $\nu_{\text{OCO}} - 1718$. ^1H NMR δ ppm: 8.83 s (8H, β -H); 8.44 d (8H, $J=8.0$ Hz, 2,6-H-Ar); 8.31 d (8H, $J=8.0$ Hz, 3,5-H-Ar); 4.53 t (8H, $^1J=6.7$ Hz, OCH_2); 1.93 sc (8H, $^1J=6.7$ Hz, CH_2); 1.25 t (12H, $^1J=6.7$ Hz, CH_3); -2.76 bs (2H, NH).

5,10,15,20-Tetrakis(4-*n*-butyloxycarbonylphenyl)porphine (25, para-, $R=-(\text{CH}_2)_4\text{CH}_3$) was obtained in a similar manner as described for 5,10,15,20-tetrakis(4-propyloxycarbonylphenyl)porphine. Yield: 26.2 %. $R_f=0.30$ (benzene). EAS λ_{\max} nm (lge): 647 (3.77); 591 (3.90); 551 (4.04); 516 (4.33); 421 (5.68). IR cm^{-1} : $\nu_{\text{OCO}} - 1718$. ^1H NMR δ ppm: 8.85 s (8H, β -H); 8.43 d (8H, $J=7.9$ Hz, 2,6-H-Ar); 8.32 d (8H, $J=7.9$ Hz, 3,5-H-Ar); 4.53 t (8H, $^1J=6.8$ Hz, OCH_2); 1.94 qv (8H, $^1J=6.8$ Hz, CH_2); 1.58 sc (8H, CH_2); 1.12 t (12H, $^1J=6.8$ Hz, CH_3); -2.78 bs (2H, NH).

5,10,15,20-Tetra(4-*n*-hexyloxycarbonylphenyl)porphine (25, para-, $R=-(\text{CH}_2)_6\text{CH}_3$). A mixture of 0.4 g (0.5 mmol) **11** (*para*), 1.0 g (7.24 mmol) potash and 2.0 mL (13.60 mmol) hexyl iodide in 20 mL DMF was refluxed for 12 h, then poured into 200 mL of water, the precipitate was filtered, washed with water and dried at 70 °C. The precipitate was dissolved in methylene chloride, filtered, evaporated and chromatographed on silica gel (L 100/250), the eluate was evaporated and the porphyrin was precipitated with methanol, filtered and dried at ambient temperature in air. Yield: 0.13 g (23.1 %). $R_f=0.31$ (benzene). EAS λ_{\max} nm (lge): 647 (3.80); 590 (3.93); 552 (4.06); 516 (4.35); 421 (5.69). IR cm^{-1} : $\nu_{\text{OCO}} - 1719$. ^1H NMR δ ppm: 8.86 s (8H, β -H); 8.48 d (8H, $J=7.9$ Hz, 2,6-H-Ar); 8.35 d (8H, $J=7.9$ Hz, 3,5-H-Ar); 4.52 t (8H, $^1J=7.0$ Hz, OCH_2); 1.96 qv (8H, $^1J=7.0$ Hz, CH_2); 1.56–1.66 m (8H, CH_2); 1.41–1.50 m (8H, CH_2); 1.19–1.43 m (8H, CH_2); 0.98 t (12H, $^1J=7.0$ Hz, CH_3); -2.78 bs (2H, NH).

5,10,15,20-Tetrakis(4-*n*-heptyloxycarbonylphenyl)porphine (25, para-, $R=-(\text{CH}_2)_7\text{CH}_3$). Yield 22.4 %. $R_f=0.48$ (benzene). EAS λ_{\max} nm (lge): 647 (3.81); 590 (3.94); 551 (4.07); 516 (4.36); 421 (5.70). IR cm^{-1} : $\nu_{\text{OCO}} - 1717$. ^1H NMR δ ppm: 8.85 s (8H, β -H); 8.49 d (8H, $J=7.9$ Hz, 2,6-H-Ar); 8.33 d (8H, $J=7.9$ Hz, 3,5-H-Ar); 4.55 t (8H, $^1J=6.7$ Hz, OCH_2); 1.93 qv (8H, $^1J=6.7$ Hz, CH_2); 1.54–1.60 m (8H, CH_2); 1.40–1.50 m (8H, CH_2); 1.20–1.45 m (16H, CH_2); 0.92 t (12H, $^1J=6.7$ Hz, CH_3); -2.78 bs (2H, NH).

5,10,15,20-Tetra(4-*n*-hexadecyloxycarbonylphenyl)porphine (25, para-, $R=-(\text{CH}_2)_{16}\text{CH}_3$). Method A. A mixture of 50 mg (0.063 mmol) of 5,10,15,20-tetrakis(4-carboxyphenyl)porphine **11** (*para*), 67 mg (0.276 mmol) of cetyl alcohol, 19 mg (0.156 mmol) of DMAP and 73 mg (0.381 mmol) of EDAC was dissolved in 15 mL of methylene chloride and refluxed with the protection by a calcium chloride tube for 3 h. The porphyrin was isolated by chromatography on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the porphyrin was precipitated with methanol. The precipitate was filtered, washed

with methanol and dried. Yield: 12.1 mg (11.4 %). Method B. A mixture of 0.2 g (0.25 mmol) of **11** (*para*), 1.0 g (7.2 mmol) of potash, 4.4 g (14.4 mmol) of hexadecyl bromide and 0.5 g of potassium iodide in 10 mL DMF was refluxed for 10 h, then diluted with 100 mL of water, the precipitate was filtered, washed with water and dried at 70 °C, dissolved in methylene chloride, filtered purified on silica gel eluting with methylene chloride, evaporated and precipitated with methanol. The precipitate was filtered, washed with methanol and dried in air at ambient temperature. Yield: 90 mg (21.3 %). $R_f=0.88$ (benzene). EAS λ_{\max} nm (lge): 646 (3.87); 590 (3.98); 550 (4.10); 516 (4.37); 420 (5.75). IR cm^{-1} : $\nu_{\text{OCO}} - 1720$. ^1H NMR δ ppm: 8.84 s (8H, β -H); 8.46 d (8H, $J=7.9$ Hz, 2,6-H-Ar); 8.31 d (8H, $J=7.9$ Hz, 3,5-H-Ar); 4.53 t (8H, $^1J=6.7$ Hz, OCH_2); 1.94 qv (8H, $^1J=6.7$ Hz, CH_2); 1.54–1.63 m (8H, CH_2); 1.43–1.51 m (8H, CH_2); 1.19–1.43 m (88H, CH_2); 0.87 t (12H, $^1J=6.7$ Hz, CH_3); -2.78 bs (2H, NH).

5,10,15,20-Tetrakis[4-(perfluoro-1,1,3-*H*-propyloxy-carbonyl)phenyl]porphine (25, para-, $R=\text{CH}_2\text{CF}_2\text{CF}_2\text{H}$). The compound was obtained similarly to the porphyrin **25** (*para*-, $R=-(\text{CH}_2)_3\text{CH}_3$), using 0.5 mL (5.5 mmol) of *n*-perfluoro-1,1,3-*H*-propyl alcohol. The reaction time was 2 days. Yield: 51 mg (64.8 %). $R_f=0.72$ (benzene). EAS λ_{\max} nm (lge): 647 (3.77); 591 (3.91); 551 (4.04); 516 (4.35); 421 (5.68). IR cm^{-1} : $\nu_{\text{OCO}} - 1734$; $\nu_{\text{CF}} - 1273$. ^1H NMR δ ppm: 8.86 s (8H, β -H); 8.51 d (8H, $J=8.2$ Hz, 2,6-H-Ar); 8.38 d (8H, $J=8.2$ Hz, 3,5-H-Ar); 6.11 tt (4H, $^1J=51.9$ Hz, $^2J=5.1$ Hz, CHF_2); 5.06t (8H, $3J=13.3$ Hz, OCH_2); -2.78s (2H, NH).

5,10,15,20-Tetrakis[4-(perfluoro-1,1,9-*H*-nonyloxy-carbonyl)phenyl]porphine (25, para-, $R=\text{CH}_2(\text{CF}_2)_7\text{CF}_2\text{H}$). Method A. A mixture of 100 mg (0.126 mmol) of 5,10,15,20-tetrakis(4-carboxyphenyl)porphine **11** (*para*), 470 mg (1.088 mmol) of *n*-perfluoro-1,1,9-*H*-nonyl alcohol, 67 mg (0.548 mmol) of DMAP and 200 mg (1.043 mmol) of EDAC was dissolved in 25 mL of anhydrous methylene chloride and stirred at 0 °C for 1.5 h, then at ambient temperature until completion of the reaction (2 days, TLC). The porphyrin was isolated by chromatography on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated and the product was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 84 mg (27.2 %). Method B. A mixture of 100 mg (0.126 mmol) of **11** (*para*) with 30 mL of methylene chloride was subjected to ultrasound for 1 h. Then 250 mg (0.579 mmol) of *n*-perfluorononyl-1,1,9-*H*-nonyl alcohol, 37 mg (0.301 mmol) of DMAP and 150 mg (0.782 mmol) of EDAC were added to the suspension and it was stirred at 0 °C for 1.5 h, then at ambient temperature until completion of the reaction (4 days, TLC). The mixture was purified on silica gel (L 100/250), eluting with methylene chloride. The eluate was evaporated, the porphyrin was precipitated with methanol. The precipitate was filtered, washed with methanol and dried. Yield: 105 mg (33.0 %). $R_f=0.80$ (benzene). EAS λ_{\max} nm (lge): 648 (3.84); 591 (3.96); 551 (4.08); 516 (4.37); 421 (5.67). IR cm^{-1} : $\nu_{\text{OCO}} - 1738$; $\nu_{\text{CF}} - 1211$. ^1H NMR δ ppm: 8.85 s (8H, β -H); 8.50 d (8H, $J=8.0$ Hz, 2,6-H-Ar); 8.38 d (8H, $J=8.0$ Hz, 2,6-H-Ar); 6.10 tt (4H, $^1J=52.0$ Hz, $^2J=5.1$ Hz, CHF_2); 5.05 t (8H, $^3J=13.3$ Hz, OCH_2); -2.79 bs (2H, NH).

5,10,15,20-Tetrakis[3-(perfluoro-1,1,9-*H*-nonyloxy-carbonyl)phenyl]porphine (25, meta-, $R=\text{CH}_2(\text{CF}_2)_7\text{CF}_2\text{H}$). was obtained similarly to **23** (*para*-, $R=\text{CH}_2(\text{CF}_2)_7\text{CF}_2\text{H}$) by the method B using porphyrin **11** (*meta*). Reaction time was 5 days. Yield: 78 mg (25.2 %). $R_f=0.77$ (benzene). EAS λ_{\max} nm (lge): 649 (3.91); 589 (3.93); 548 (3.99); 515 (4.37); 420 (5.72). IR cm^{-1} : $\nu_{\text{OCO}} - 1741$; $\nu_{\text{CF}} - 1210$. ^1H NMR δ ppm: 9.28 s (4H, 2-H-Ar); 8.81–8.88 m (4H, 6-H-Ar); 8.73 s; 8.75 s (12H, 4-H-Ar+ β -H); 8.21 t (4H, $J=7.5$ Hz; 5-H-Ar); 6.05 t (4H, $^1J=52$ Hz; CHF_2); 5.10 t (8H, $^2J=16.4$ Hz, OCH_2); -0.76 bs (4H, NH) (CDCl_3 -TFA).

Acknowledgments. The current research was conducted in accordance with the State assignment of the Ministry of Science of Russian Federation. The porphyrinic compounds

were synthesized with the support of Russian Foundation for Basic Research, project number 14-23-00204.

References

- Salnikova M.A., Lubimova T.V., Glazunov A.V., Syrбу S.A., Semeikin A.S., Koifman O.I. *Macroheterocycles* **2013**, 6, 53–58.
- Salnikova M.A., Lubimova T.V., Glazunov A.V., Syrбу S.A., Semeikin A.S. *Macroheterocycles* **2014**, 7, 249–255.
- Salnikova M.A., Prigorelov G.A., Syrбу S.A., Golubchikov O.A., Semeikin A.S. *Izv. Vyssh. Uchebn. Zaved., Khim. Khim. Tekhnol.* **2013**, 56, 38–41 (in Russ.).
- Li Y., Cao L., Tian H. *J. Org. Chem.* **2006**, 71, 8279–8282.
- Maligaspe E., Tkachenko N.V., Subbaiyan N.K., et al. *J. Phys. Chem. A* **2009**, 113, 8478–8489.
- Sobral A.J.F.N., Melo S.M., Ramos M.L., Teixeira R., Andrade S.M., Costa S.M.B. *Tetrahedron Lett.* **2007**, 48, 3145–3149.
- Wu Z.-Q., Shao X.-B., Li C., Hou J.-L., Wang K., Jiang X.-K., Li Z.-T. *J. Am. Chem. Soc.* **2005**, 127, 17460–17468.
- Bakleh M.E., Sol V., Estieu-Gionnet K. *Tetrahedron* **2009**, 65, 7385–7392.
- Xiao Z.-Y., Hou J.-L., Jiang X.-K. *Tetrahedron* **2009**, 65, 10182–10191.
- Stasio B.D., Frochot C., Dumas D., Even P., Zwier J., Muller A., Didelon J., Guillemin F., Viriot M.-L., Barberi-Heyob M. *Eur. J. Med. Chem.* **2005**, 40, 1111–1122.
- Lebedev A.Y., Cheprakov A.V., Sakadžić S., Boas D.A., Wilson D.F., Vinogradov S.A. *Appl. Mater. Interfaces* **2009**, 1, 1292–1304.
- Burrell A.K., Officer D.L., Plieger P.G., Reid D.C.W. *Chem. Rev.* **2001**, 101, 2751–2796.
- Matsumoto J., Matsumoto T., Senda Y., Shiragami T., Yasuda M. *J. Photochem. Photobiol., A* **2008**, 197, 101–109.
- Han Z.-X., Luo H.-Y., Zhang X.-B., et al. *Spectrochim. Acta: Part A* **2009**, 1084–1088.
- Bakar M.B., Oelgemoller M., Senge M.O. *Tetrahedron* **2009**, 65, 7064–7078.
- Kolodina E.A., Lubimova T.V., Syrбу S.A., Semeikin A.S. *Macroheterocycles* **2009**, 2, 33–41.
- Kolodina E.A., Syrбу S.A., Semeikin A.S., Koifman O.I. *Zh. Org. Khim.* **2010**, 46, 136–141 (in Russ.).
- Mikhalitsyna E.A., Tyurin V.S., Nefedov S.E., Syrбу S.A., Semeikin A.S., Koifman O.I., Beletskaya I.P. *Eur. J. Inorg. Chem.* **2012**, 5979–5990.
- In: *Organikum* (transl. from Germ.), Moscow: BINOM, **2008**, 2, p. 361–392 (in Russ.) [В кн.: Органикум (пер. с нем.), М.: Бином, 2008, с. 361–392].

Received 20.10.2015

Revised 15.12.2015

Accepted 06.01.2016